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Modern Chemistry Chapter 14. boiling-point elevation. colligative properties. dissociation. freezing-point depression. the difference between the boiling point of a pure... solvent and.... properties that depend on the concentration of solute... particle.... the separation of ions that occurs when an ionic compound... diss....

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Modern Chemistry 127 Chapter Test Name Class Date Chapter Test A, continued ____14. A strong base in an aqueous solution a. is a weak electrolyte. b. produces many H ions. c. will not dissolve. d. completely dissociates into ions. ____15. In a Brønsted-Lowry acid-base reaction, what are transferred from one reactant to another? a. electrons ...

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Particle Mass number 19. Proton 20. Neutron 21. Electron 10 CHAPTER 3 TEST Relative charge Location MODERN CHEMISTRY HRW material copyrighted under notice appearing earlier in this work. Menu Print Name Date Class CHAPTER 3 TEST continued SHORT ANSWER Write the answers to the following questions in the space provided. 22.

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Modern Chemistry 33 Chapter Test Name Class Date Chapter Test B, continued 15. The energy state of an atom is called its ground state. 16. The number of waves that pass a point in one second is called. 17. When an electron drops from a higher-energy state to a lower-energy state, a(n) spectrum is produced. 18.

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