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2.2 MASS-ACTION KINETICS 2.2.1

OVERVIEW We assume that rates can be described by $\text{Rate} = (k) \cdot (\text{rate law})$ where (k) is the rate constant from Arrhenius' law and (rate law) is the rate law. Empirically, we say that $(\text{rate law}) = \dots$ which is typical rate law kinetics.

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